

VALENCE BAND STUDY OF $\text{LaNiO}_{3-\delta}$ THIN FILMS

S. Grebinskij¹, M. Senulis¹, H. Tvardauskas¹, V. Bondarenka¹, V. Lisauskas¹, K. Sliuziene¹,
B. Vengalis¹, B.A. Orlowski², R.L. Johnson³, and S. Mickevicius^{1*}

¹ Institute of Semiconductor Physics, A. Gostauto 11, LT-01108 Vilnius, Lithuania

² Institute of Physics, Polish Academy of Sciences, Al. Lotnikow 32/46, 02-668 Warsaw, Poland

³ Hamburger Synchrotronstrahlungslabor HASYLAB am Deutschen Elektronen-Synchrotron DESY, Notkestr. 85, D-22603 Hamburg, Germany

Keywords: rare earth alloys and compounds, resonance photoemission, valence band electronic structure, oxide materials

*) e-mail: sigism@pfi.lt

LaNiO_3 is one of the few conductive oxides with a crystal structure suitable for integration in epitaxial heterostructures with perovskites of enormous technological potential such as colossal magnetoresistance materials, high-temperature superconductors and ferroelectrics. It is known that the considerable surface segregation of elements may place in LaNiO_{3-x} samples and is the tendency of rare earth and nickel oxides to absorb water vapor and carbon dioxide from air, so that any *ex situ* exposure of these films to air will result in an uncontrolled reaction and surface stoichiometry variation [1-2]. Thus the knowledge of the surface composition is extremely important because it is directly related to the heterostructures properties.

The initially hydrated $\text{LaNiO}_{3-\delta}$ surface may be restored by heating above dehydration temperature. Nickel hydroxide, in turn, decomposes at $T > 230^\circ\text{C}$ (melting point). When heated to decomposition it emits toxic fumes of metallic nickel, and one would expect a decrease of Ni-species relative concentration in the previously hydrated surface layer. The aim of this work is to investigate the valence band electronic structure and chemical composition of $\text{LaNiO}_{3-\delta}$ thin films after heating above dehydration temperature about 500°C .

Thin LaNiO_{3-x} films onto monocrystalline (100)-plane oriented NdGaO_3 substrate were deposited by using a reactive DC magnetron sputtering technique. To prevent the film bombardment by high energy ions during deposition, NdGaO_3 substrates were positioned in "off-axis" configuration at a distance of 15 mm from the symmetry axis of the discharge and 20 mm over the target plane. The substrate temperature was $\sim 750^\circ\text{C}$ and the resultant thickness of LaNiO_{3-x} film was about $0.1 \mu\text{m}$.

The resonant photoemission experiments were performed in the synchrotron radiation laboratory

HASYLAB, Hamburg (Germany). Synchrotron radiation obtained from the storage ring DORIS III was monochromatized with the FLIPPER II plane grating vacuum monochromator designed for the photon energy range of 15–200 eV. The spectrometer was equipped with a CMA electron energy analyzer. The total energy resolution was kept at 0.1 eV. The origin of the energy axis was set at the Fermi energy as measured for a reference metallic sample.

The giant resonance in La 5*p* and La 5*s* peaks intensity observed at excitation energy corresponding to a La[4*d* → 4*f*] threshold ($h\nu = 119.5 \text{ eV}$) is accompanied by a weak resonance of $\text{N}_{4,5}\text{O}_{2,3}\text{O}_{2,3}$ and $\text{N}_{4,5}\text{O}_{2,3}\text{V}$ Auger peaks. The obtained results are in an agreement with the model of an autoionization process after resonant excitation. The relatively weak enhancement of the intensity of valence band maxima (at about 6 eV) may be explained by the small mixing of the La 5*d* ionic character to the O 2*p* valence band. No resonant features were observed in the VB spectra under Ni[3*p* → 3*d*] excitation (escape depth $L \approx 2 \text{ ML}$), indicating that nickel species are not presented at the $\text{LaNiO}_{3-\delta}$ film surface after heat treatment.

References

- [1] J. Choisnet, N. Abadzhieva, P. Stefanov, D. Klissurski, J.M. Bassat, V. Rives, L. Minchev, "X-ray photoelectron spectroscopy, temperature-programmed desorption and temperature-programmed reduction study of LaNiO_3 and $\text{La}_2\text{NiO}_{4+\delta}$ catalysts for methanol oxidation", *J. Chem. Soc. Faraday Trans.* **90** (1994) 987–991.
- [2] Y. Li, N. Chen, J. Zhou, S. Song, L. Liu, Z. Yin, C. Cai, "Effect of the oxygen concentration on the properties of Gd_2O_3 thin films", *J. Cryst. Growth* **265** (2004) 548–552.